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SYNTHETIC STUDIES ON SICCANIN. EFFICIENT CONSTRUCTION OF THE CIS-FUSED DRIMANE UNIT AND SYNTHESIS OF ISOSICCANIN METHYL ETHER

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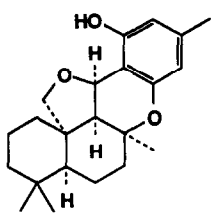
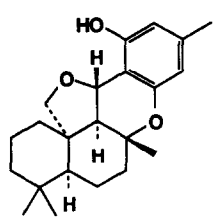
Abstract: Under ferric chloride catalysis, the Diels-Alder addition of dienone ester **3** to diene **4** occurred readily to give adduct **5** which possesses the complete carbon framework of the sesquiterpene unit of siccanin (**1**) with the correct relative stereochemistry. Further elaboration on **5** gave rise to the methyl ether of isosiccanin (**2**), an unnatural diastereomer of the antifungal antibiotic **1**.

Siccanin was first isolated by Ishibashi from the culture broth of *Helminthosporium siccans* Drechsler.[1] As shown in formula **1**, this compound possesses a unique structure consisting of an unusual *cis*-fused drimane sesquiterpene unit and an orsellinate moiety.[2] Biologically, it is a highly potent, yet rather nontoxic antifungal antibiotic[3] which is particularly effective against various human pathogenic fungi. Owing to these interesting properties, siccanin (**1**) and its analogues have been the subject of considerable synthetic activity[3-5] with an elegant total synthesis of the natural metabolite having been recently accomplished.[5] Herein we wish to describe a facile approach to the *cis*-fused drimane system with excellent regio- and stereochemical control as well as the synthesis of the methyl ether of an unnatural diastereomer of **1** which we name isosiccanin (**2**).



1



2

The construction of the sesquiterpene portion of siccanin (**1**) was facilitated by the observed dienophilicity of enone ester **3**. This compound was shown to undergo Diels-Alder reaction with a variety of 1,3-butadienes resulting in the direct formation of highly functionalized decalin systems with expectable regio- and stereochemistry.[6] Based on these findings, the

addition of **3** to diene **4** is expected to give keto ester 5 possessing the complete carbon framework of the desired sesquiterpene unit. Indeed, when enone ester **3** was subjected to Diels-Alder reaction with diene **4**[7] (1.5 eq) in ether under the catalysis of ferric chloride (1 eq), the desired adduct **5** was isolated in 84% yield after 48 h at -20°C (Scheme 1). The structure of this compound could be readily deduced on the basis of the spectral data and the observed mode of cycloaddition of **3** with various dienes.[6] This assignment was unambiguously confirmed at a later stage by X-ray analysis of a derivative.

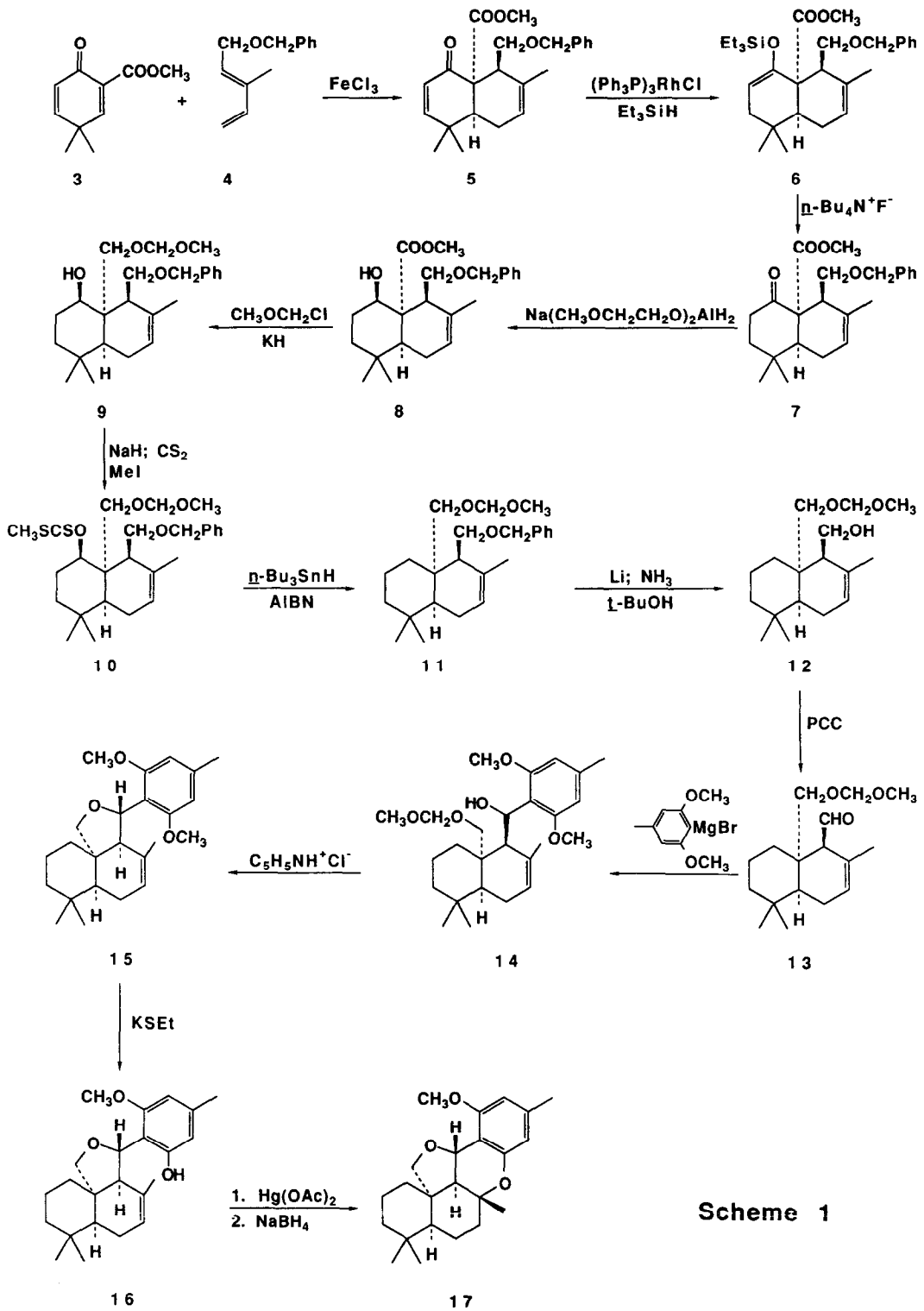
In order to selectively saturate the conjugated carbon-carbon double bond, compound 5 was treated with triethylsilane and a catalytic amount of tris(triphenylphosphine)rhodium(I) chloride in benzene[9] followed by exposure of the resulting silyl enol ether **6** to a solution of tetra-n-butylammonium fluoride in tetrahydrofuran. Keto ester **7** thus obtained in quantitative yield was reduced with sodium bis(2-methoxyethoxy)aluminum hydride at room temperature in ether and toluene giving diol **8** (m.p. 90-92°C; 72% yield), the structure of which was unambiguously determined by a single crystal X-ray analysis.

To remove the successive secondary hydroxyl group, diol **8** was converted to hydroxy ether **9** (72% yield) by treatment with an excess of potassium hydride and a stoichiometric amount of chloromethyl methyl ether at -10°C using 1,2-dimethoxyethane as a solvent. Deoxygenation of **9** was effected via the corresponding xanthate **10** which was readily formed when the former was treated with carbon disulfide, methyl iodide and sodium hydride at room temperature in 1,2-dimethoxyethane. Heating xanthate **10** with tri-n-butyltin hydride[10] and a small amount of azobisisobutyronitrile in degassed toluene at reflux gave the desired ether **11** (93% yield from **9**).

Debenzylation of **11** using lithium in liquid ammonia containing a small quantity of t-butyl alcohol gave rise to alcohol **12**, which was oxidized to the corresponding aldehyde **13** (66% yield from **11**) on exposure to pyridinium chlorochromate[11] in dichloromethane. At this point, the construction of the complete sesquiterpene unit present in the target molecule with the correct stereochemistry and oxidation level has been successfully accomplished in nine steps with 27% overall yield from dienone ester **3**.

The orsellinate moiety was incorporated by the reaction of aldehyde **13** with 2,6-dimethoxy-4-methylmagnesium bromide carried out in ether at a temperature range of -78°C to 20°C. A single stereoisomer **14**, with the undetermined stereochemistry at the newly generated chiral center, was produced in 70% yield.

The transformation of **14** to siccanin **1** requires consecutive closures of two ether rings. On exposure to pyridinium chloride in refluxing dichloromethane, the tetrahydrofuran ring was readily formed and a single product, m.p. 179-181°C, was isolated in quantitative yield. Unfortunately, a single


COOCH3
CH2OCH2Ph
FeCl3
(Ph3P)3RhCl
Et3SiH
Et3SiO
n-Bu4N+F-
CH2OCH2OCH3
HO
CH3OCH2Cl
KH
Na(CH3OCH2CH2O)2AlH2
NaH; CS2
MeI
CH3SCSO
n-Bu3SnH
AIBN
Li; NH3
t-BuOH
CH2OH
PCC
CHO
OCH3
MgBr
C5H5NH+Cl-
KSEt
1. Hg(OAc)2
2. NaBH4
3
4
5
6
7
8
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10
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14
15
16
17

Scheme 1

crystal X-ray analysis revealed that this compound was the epimer **15**. This compound, which is expected to be thermodynamically more stable than the desired epimer, possesses an incorrect center for **1**. Its demethylation using potassium ethanethiolate[12] in hot dimethylformamide followed by sequential treatment of the resulting phenol **16** with mercuric acetate and sodium borohydride in methanol at room temperature gave isosiccanin methyl ether **17** (48% yield), m.p. 208-209°C, which displayed spectral data [ir ($CHCl_3$ cast) 1616 (Ar-O-R) and 1114 cm^{-1} (R-O-R); nmr ($CDCl_3$) δ 6.32, 6.24 (both s, 1H each, aromatic), 5.56 (br.s, 1H, ArCHO-), 4.16 (d, 1H, J = 8 Hz, -OCHH-), 3.80 (s, 3H, $-OCH_3$), 3.56 (dd, 1H, J = 8, J' = 1 Hz, -OCHH-), 2.28 (s, 3H, $ArCH_3$), 1.50 (s, 3H, CH_3CO-), 0.84 and 0.82 (both s, 3H each, $-C(CH_3)_2$)] similar but not identical to those reported[2c] for the methyl ether derived from the naturally occurring compound **1**.

Redirection of the latter part of the approach towards the synthesis of siccanin (**1**) is under current investigation.[13]

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